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**ANTIPLASMIN THE MAIN PLASMIN INHIBITOR IN BLOOD
PLASMA**

Studies on Structure-Function Relationships

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ABSTRACT

ANTIPLASMIN THE MAIN PLASMIN INHIBITOR IN BLOOD PLASMA***Studies on Structure-Function Relationships*****Haiyao Wang****Department of Surgical Sciences, Division of Clinical Chemistry and Blood Coagulation, Karolinska University Hospital, Karolinska Institute, S-171 76 Stockholm, Sweden**

Antiplasmin is an important regulator of the fibrinolytic system. It inactivates plasmin very rapidly. The reaction between plasmin and antiplasmin occurs in several steps: first a lysine-binding site in plasmin interacts with a complementary site in antiplasmin. Then, an interaction occurs between the substrate-binding pocket in the plasmin active site and the scissile peptide bond in the RCL of antiplasmin. Subsequently, peptide bond cleavage occurs and a stable acyl-enzyme complex is formed. It has been accepted that the COOH-terminal lysine residue in antiplasmin is responsible for its interaction with the plasmin lysine-binding sites. In order to identify these structures, we constructed single-site mutants of charged amino acids in the COOH-terminal portion of antiplasmin. We found that modification of the COOH-terminal residue, Lys452, did not change the activity or the kinetic properties significantly, suggesting that Lys452 is not involved in the lysine-binding site mediated interaction between plasmin and antiplasmin. On the other hand, modification of Lys436 to Glu decreased the reaction rate significantly, suggesting this residue to have a key function in this interaction. Results from computerised molecular modelling indeed supported our experimental data. The interaction between immobilized plasminogen or an elastase degradation product from plasminogen, constituting “kringles” 1-3 and different purified variants of antiplasmin was then studied by surface plasmon resonance. Again, the data demonstrated that Lys452 is not involved in the lysine-binding site mediated interaction. On the other hand, solid evidence was produced, proving that Lys436 is indeed very important for this interaction. Some evidence was found that Glu443 may also be involved in this interaction. Molecular modelling experiments suggest that the negatively charged Glu443 is within the expected range from the positively charged Lys436 to form a complementary site to a lysine-binding site. Serine protease inhibitors may under certain conditions undergo conformational changes resulting in the insertion of RCL into the A β -sheet during formation of “latent” molecules or polymers. Antiplasmin is stable at neutral pH, but at acidic pH or at elevated temperatures it rapidly becomes inactivated. At decreased pH, antiplasmin activity declined following first-order kinetics. Analysis by PAGE under non-denaturing conditions demonstrated that only minor amounts of polymerized material had formed. However, on incubation at elevated temperatures a rapid formation of polymerized material was observed. Antiplasmin inactivated by treatment at pH ~5 could spontaneously regain activity if incubated at neutral pH. Furthermore, by treatment of such material with guanidinium chloride followed by dialysis, considerable activity was regained, in contrast to antiplasmin that had been inactivated by polymerization. To better understand these processes, site-directed mutagenesis was employed to produce some interesting variants of antiplasmin, which were purified and characterized. Five of the 11 mutants were found to have a deviating stability at decreased pH. One mutant was less stable as compared to wt-antiplasmin, but the other 4 were more stable. His341Thr was 7-fold more stable at pH 4.9, as compared to wt-antiplasmin. The wt-antiplasmin had a much more pronounced tendency to polymerize at decreased pH, as compared to “native” antiplasmin. However, many of the mutants were rather transformed to “latent” molecules, as judged both from PAGE-analysis at non-denaturing condition and reactivation experiments.

Key words: antiplasmin, lysine-binding-site, inactivation.**ISBN 91-7140-278-0**

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TABLE OF CONTENTS

LIST OF PUBLICATIONS	6
ABBREVIATIONS	7
INTRODUCTION	8
THE COMPOUNDS OF THE FIBRINOLYTIC SYSTEM	9
Plasminogen and Plasmin	9
t-PA and u-PA	9
Antiplasmin	10
Plasminogen activator inhibitor-1 (PAI-1)	10
Plasminogen activator inhibitor-2 (PAI-2)	10
Thrombin-activatable fibrinolysis inhibitor (TAFI)	11
SERPINS	11
General introduction about serpins	11
The structural properties of serpins	11
The interaction of serpins and proteases	11
Conformational changes: cleaved, “latent” and polymerisation	12
Antiplasmin	13
PAI-1	14
Antithrombin	15
Conformational disease	16
THE PRESENT INVESTIGATIONS	16
Aims of the study	16
Methodology	16
Site-directed mutagenesis of antiplasmin cDNA	16
Expression of antiplasmin variants in S2 cell	17
Purification of antiplasmin variants	17

Determination of antiplasmin activity and antigen concentration	18
Polyacrylamide gel electrophoresis (PAGE)	18
Determination of rate constants in the reaction between plasmin and the antiplasmin variants	19
Binding of antiplasmin variants to plasminogen or “kringles” 1-3 measured by surface plasmon resonance	19
Computer modelling of the COOH-terminal 25 or 40 amino acid residues in antiplasmin	20
RESULTS AND DISCUSSION	20
The site in antiplasmin interacting with lysine-binding sites in plasmin (Papers I and II)	20
The reactions between plasmin and different antiplasmin variants	20
The Interaction between Plasminogen and Antiplasmin Variants as Studied by Surface Plasmon Resonance	21
Molecular modelling of the COOHterminal portion of antiplasmin	21
Antiplasmin stability (Papers III and IV)	22
Inactivation of antiplasmin	22
Reactivation of antiplasmin	23
Structures of importance for the stability of antiplasmin as studied by site-directed mutagenesis	23
ACKNOWLEDGEMENTS	26
REFERENCES	27

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- III. Wang H, Pap S and Wiman B. (2004) Inactivation of antiplasmin at low pH: evidence for the formation of latent molecules, *Thrombosis Research*, 114: 301-306

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ABBREVIATIONS

AP	antiplasmin
PAGE	polyacrylamide gel electrophoresis
PAI-1	plasminogen activator inhibitor-1
PAI-2	plasminogen activator inhibitor-2
RCL	reactive centre loop
serpin	serine protease inhibitor
SDS	sodium dodecyl sulphate
tPA	tissue-type plasminogen activator
uPA	urokinase
6-AHA	6-aminohexanoic acid
wt-AP	wild type antiplasmin

INTRODUCTION

The fibrinolytic system is capable of dissolving blood clots and removing excessive fibrin from the blood vessels. This system is also involved in tissue repair, ovulation, macrophage function and malignant transformation. The main components of the fibrinolytic system are serine proteases, such as plasmin (Wallen and Wiman, 1970, 1972; Wiman and Wallen, 1973, 1975a 1975b, 1975c), tissue-type plasminogen activator (t-PA) (Rijken and Collen, 1981; Pennica et al, 1983; Bykowskan et al, 1981; Pannekoek et al, 1988) or urokinase-type plasminogen activator (u-PA) (Wun et al, 1982; Nielsen et al, 1982). Serine protease inhibitors (Serpins) like antiplasmin (Collen, 1976; Moroi and Aoki, 1976; Mullertz and Clemmensen, 1976) and plasminogen activator inhibitor type-1 (PAI-1) (Loskutoff et al, 1983; Chmielewska et al, 1983; Kruithof et al, 1984) are important regulators of this system. The proteolytic enzyme plasmin is formed by activation of plasminogen by t-PA or u-PA. In the blood

vessels, this occurs preferably at the fibrin surface, since both plasminogen and t-PA have a high affinity for fibrin. This is one important way to keep the fibrinolytic system localized to the fibrin surface. When fibrin has become degraded into soluble degradation products, free plasmin will rapidly become inactivated by antiplasmin (Wiman and Collen, 1977). This is another regulating step, also keeping the fibrinolytic process localized. In addition the plasminogen activator inhibitors (in plasma mainly PAI-1) regulate the activity of the activators.

Increased fibrinolytic activity, e.g. due to antiplasmin deficiency, is associated with a bleeding tendency. A decreased fibrinolytic activity, on the other hand, is often associated with thromboembolic diseases. Plasminogen activators like t-PA and u-PA are used as thrombolytic agents in the treatment of myocardial infarction and in other thrombotic diseases.

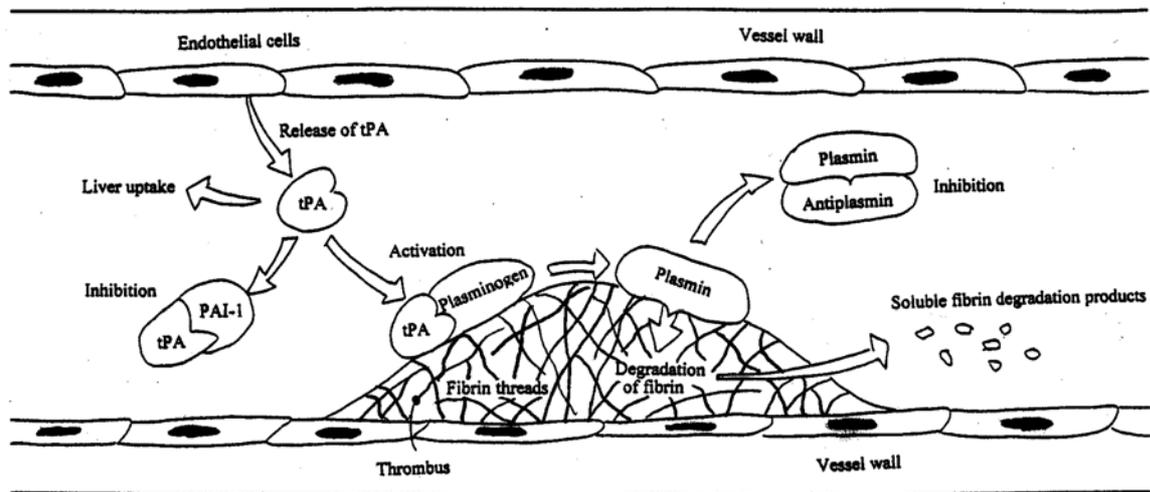


Figure 1. A schematic picture of the fibrinolytic system (adapted from Wiman, MFR informerar 1987)

THE COMPOUNDS OF THE FIBRINOLYTIC SYSTEM

Plasminogen and Plasmin

Human plasminogen (EC 3.4.21.7) is a single-chain glycoprotein of 92 kd and contains approximately 2% carbohydrate (Collen and Maeyer, 1975). The gene for human plasminogen is located on chromosome 6q26-q27; it spans 53.5 kilobases (kb) and is the largest gene among the constituents of the fibrinolytic system (Magnaghi et al, 1994). The molecule consists of 791 amino acids and contains 24 disulfide bonds. It has a concentration of about 1.5 μ M in plasma (Ries, 1997). Plasminogen contains five kringles on the A-chain. Each of the five kringles consists of 78 to 80 amino acid residues. The kringle is important in creating the ability for plasminogen to bind to exposed lysyl residues in fibrin (Wiman and Wallen, 1977a; Thorsen et al, 1981; Thewes et al, 1990), in antiplasmin (Wiman et al, 1979a), in tetranectin (Clemmensen et al, 1986), in histidine-rich glycoprotein (Lijnen et al, 1980), in collagen (Kelm et al, 1994), in high-molecular-weight and low-molecular-weight kininogen (Selim et al, 1997), in extra cellular matrix (Kundsen et al, 1986), in immunoglobulin G (Harpel et al, 1989) or in cell surface receptors (Miles et al, 1991; Hajjar et al, 1986; Ullberg et al, 1989). The catalytic domain of plasminogen (B chain) is made up of 230 amino acids. The active site consists of the three amino acids Ser741, His603 and Asp646. Plasminogen can exist in two forms: glu-plasminogen and lys-plasminogen. The glu-plasminogen is the natural form, which can be cleaved at the Arg68-Met69, Lys77-Lys78 or Lys78-Val79 to form lys-plasminogen. Lys-plasminogen is more readily convertible to plasmin than glu-plasminogen (Wallen and Wiman, 1970, 1972; Wiman and Wallen, 1975a, Wiman and Collen, 1978c; Collen and Maeyer, 1975).

The conversion of plasminogen to plasmin is conducted by t-PA or u-PA, by cleaving the

peptide bond Arg561-Val562 in plasminogen and thereby generates active two-chain plasmin (Ling et al, 1965).

t-PA and u-PA

One of the two-plasminogen activators in humans is t-PA, a serine protease (EC 3.4.21.68) of 68 kd. The t-PA gene is located on bands p12-p11 on chromosome 8 (Benham et al, 1984). In normal plasma, the antigen concentration of t-PA is about 10 μ g/L (Rijken and Collen, 1981). Most of the t-PA in plasma is present in a complex with its primary inhibitor, PAI-1 (Chmielewska et al, 1983) or to some of the other protease inhibitors in plasma (Thorsen and Philips, 1984; Nordenhem and Wiman, 1998). t-PA is mainly expressed in the endothelial cells of the blood vessels. The single-chain t-PA is unique among serine proteases, since it is almost fully active, especially in the presence of fibrin (Wallen et al, 1982). However, it can be converted to two-chain t-PA by cleaving the peptide bond Arg275-Ile276 (Canipari and Strickland, 1985; Saksela and Rifkin, 1988), thereby increasing its activity, especially in the absence of fibrin (Wallen et al, 1983). The t-PA structure contains a finger domain, an epidermal growth factor domain, two kringle domains and a catalytic domain. The tPA-molecule can bind to the surface of fibrin, mainly via kringle 2 and this binding stimulates the activity of t-PA (Kelley et al, 1991). The catalytic domain of t-PA consists of 230 amino acids whose sequence is homologous to that of the other serine proteases. The active site contains the three amino acids: His322, Asp371 and Ser478 (Hoylelaerts et al, 1982; Kaneko et al, 1992; Berg et al, 1993; Collen and Lijnen, 1991). Recombinant variant of t-PA is commonly used as thrombolytic agents, especially in connection with the acute phase of myocardial infarction.

Urokinase, or u-PA, (EC 3.4.2.73) is found in relatively high concentration in human urine. It is secreted by various cell types as a single-chain glycoprotein with a molecular mass of 55 kd (Wun et al, 1982; Nielsen et al, 1982). The human u-PA gene is 6.4 kb long

and situated on chromosome 10q24 (Triputti et al, 1985). Like t-PA, it belongs to the serine protease family. Three different domains can be distinguished on the basis of the amino acid sequence. The epidermal growth factor domain interacts with the u-PA receptor. The single kringle domain has no affinity for fibrin. The protease domain contains the His204, Asp255 and Ser356, which form the catalytic triad of the active centre. The principal cleavage site that converts the inactive single chain molecule to the active two-chain form is Lys158-Ile159. As a clinical thrombolytic agent, single-chain u-PA can be used together with t-PA to get thrombolysis (Ichinose et al, 1986; Vassalli et al, 1991).

In tissues, u-PA plays an important role in the degradation of extra cellular matrix that enables cells to migrate. It also plays important role in wound healing, inflammation, embryogenesis, and invasion of tumour cells and metastasization of tumours (Dano et al, 1985; Gyetko et al, 1994; Ossowski, 1988, Ossowski et al, 1991). Thus, it activates plasminogen to plasmin, which in turn may activate matrix metalloproteases. These then degrades the extracellular matrix.

Antiplasmin

Antiplasmin, also called plasmin inhibitor, is the main inhibitor of plasmin in human plasma. It is a single chain 70 kd glycoprotein and its concentration in plasma is about 70 mg/L (Moroi and Aoki, 1976; Collen, 1976). It is synthesized in the liver and reduced in patients with liver disease. Antiplasmin is an important regulator in the fibrinolytic system. It inhibits plasmin rapidly, interferes with the adsorption plasmin(ogen) to fibrin (Collen and Wiman, 1979; Lijnen et al, 1987). It belongs to serpin super family of proteins. Deficiency of antiplasmin leads to a severe bleeding disorder (Sakata and Aoki, 1982). For a more detailed description of antiplasmin structure and function, see below (Serpine chapter).

Plasminogen activator inhibitor-1 (PAI-1)

There are several plasminogen activator inhibitors in plasma, PAI-1, PAI-2 (placental plasminogen activator inhibitor), PAI-3 (protein C inhibitor) and protease nexin (Kruithof, 1988). PAI-1, a 52-kd glycoprotein, is the most important inhibitor of the plasminogen activators (Loskutoff et al, 1983; Chmielewska et al, 1983; Kruithof et al, 1984). A few individuals with total PAI-1 deficiency have been described. They seem to suffer from a mild bleeding tendency (Stankiewicz et al, 1991). An increased plasma level of PAI-1, on the other hand, is connected to a thrombotic tendency, both venous thromboembolism (Paramo et al, 1985; Eriksson et al, 1989; Schulman and Wiman, 1996) and myocardial infarction (Hamsten et al, 1987; Held et al, 1997; Cortellaro et al, 1993). For a more detailed description of PAI-1 structure and function see below (Serpine chapter).

Plasminogen activator inhibitor-2 (PAI-2)

PAI-2 was discovered in human placenta (kawano et al, 1968), later on it was found in macrophages and several other different cell lines. In normal plasma, it is typically not found. However, during pregnancy high levels of PAI-2 are found also in plasma (Kruithof et al, 1987; Lecander and Åstedt 1987). Also in many patients with acute myeloblastic leukaemia PAI-2 could be found in high concentrations (Scherrer et al, 1991). PAI-2 is a single chain protein, consisting of 425 amino acids (Ye et al, 1987; Ny et al, 1989). The gene of PAI-2 spans 16.5 kb and is located on chromosome 18q22.1 (Ye et al, 1987, 1989). PAI-2 is an efficient inhibitor of t-PA (Kruithof et al, 1986). It also inhibits tct-PA, but it is a poor inhibitor of sct-PA (Kruithof et al, 1986). The exact physiological role of PAI-2 is not known, but an impaired fibrinolysis could be important in late state pregnancy or even more so during delivery.

Thrombin-activatable fibrinolysis inhibitor (TAFI)

TAFI (EC 3.4.17.20) is a 60 kd glycoprotein of 417 amino acids, typically synthesised in the liver. It is also known as procarboxypeptidase U or plasma procarboxypeptidase B. It can be activated by thrombin, trypsin, kallikrein or plasmin into the active enzyme TAFIa (carboxypeptidase U or plasma carboxypeptidase B). The most efficient activator seems to be the complex of thrombin and thrombomodulin (Redlitz et al, 1995; Bajzar et al, 1995, 1996). Activation occurs by releasing a 92-residue activation peptide and generates the active 35-kd enzyme. TAFIa can potentially inhibit fibrinolysis by removing carboxyterminal lysine residues from partially degraded fibrin, thereby decreasing plasminogen binding to the surface of fibrin (Sakharov et al, 1997; Wang et al, 1998). TAFI is therefore not an inhibitor but an enzyme that may modify fibrinolytic activity.

SERPINS

General introduction about serpins

Serpins (serine protease inhibitors) form a super family of proteins including a large number of members of different species. The membership is based on the presence of a single common core domain consisting of three β -sheets and 8-9 α -helices. Serpins contain more than 40 members that display more than 35% of homology regarding their primary structures. These inhibitors were identified from viruses, insects, plants, animals and human (Gettins, 2002). Because of their importance in regulation of many physiological processes, such as blood coagulation, fibrinolysis and inflammation, human serpins are the best characterized. Of the 34 known human serpins, at least seven have no inhibitory function. Some of the functional serpins are α_1 -protease inhibitor, PAI-1, antithrombin and antiplasmin, whereas some non-inhibitory members are

ovalbumin or maspin. Each inhibitory serpin has its own specific target proteases.

The structural properties of serpins

The first crystal structure of a serpin was obtained in 1984 and was that of cleaved human α_1 -protease inhibitor (Leobermann et al, 1984). Later on several other serpins were crystallized and their structures determined by X-ray crystallography. Thus, data were reported for ovalbumin (Stein et al, 1990, 1991), α_1 -antichymotrypsin (Baumann et al, 1991), PAI-1 (Mottonen et al, 1992) and antithrombin (Mourey et al, 1993). They all share similarities in their three-dimensional structures, such as three dominant β -sheets (A, B and C), nine α -helices (A - I) and one RCL (RCL) (Huber and Carrell, 1989; Baumann et al, 1991). A β -sheet is the largest of the three β -sheets and is composed of five strands, the first strand is a short strand of 5-6 residues and the remainder are longer strands of similar length, about 12-15 residues. The central strands of the A β -sheet has a parallel direction, but the others are antiparallel. The B β - and C β -sheets are shorter and composed of four-six strands. In the native serpin conformation, the RCL is well exposed at the surface of the protein. This loop contains the scissible peptide bond in those serpins with an inhibitory function and therefore interacts with the active site of the target proteases. The length of the RCL is typically about 17 residues with very little variation (Stein et al, 1990).

The interaction of serpins and proteases

The typical reaction between serpins (I) and proteases (E) is schematically demonstrated in Fig 2. The first stage is the reversible formation of a non-covalent complex (E•I). At this stage, the serpin molecule is still intact. The second stage involves a proteolytic attack on the scissile peptide bond, resulting first in the formation of a tetrahedral intermediate (O'Malley et al, 1997) and later on in a complex where the P1-P1' peptide bond is cleaved (Wiman and Collen, 1978a; Olson and Shore, 1982; Lawrence et al, 1995;

Wilczynska et al, 1995; Egelund et al, 1998). After the formation of E•I, there are two possible pathways for the intermediate. Firstly, in the inhibitor pathway, cleavage of the peptide bond, P1-P1', in the serpin occurs and is followed by the formation of a covalently stabilized complex (E-I*). Secondly in the substrate pathway, the formed complex is hydrolyzed, forming a cleaved inhibitor (I*) and release of the free enzyme (E).

From the structural point of view, serpins undergo major conformational changes upon binding of proteases (Gettins et al, 1993). The RCL of a serpin is cleaved in the final complex (Nilsson and Wiman, 1982). P1' and P3 residues are separated by a distance of 60 Å in the complex between the serpin and the protease (Wilczynska et al, 1997). Therefore, after cleavage of the scissile peptide bond in RCL part of the RCL rapidly becomes inserted into the A β-sheet of the serpin (Fa et al, 1995; Shore et al, 1995; Stratikos and Gettins, 1997, 1998). This results in that the covalently attached protease is dragged from the initial docking position on the serpin to the opposite side. This destroys the active site of the enzyme and entraps the enzyme in an inactive complex with the inhibitor (Wilczynska et al, 1997). The progression of the inhibitor pathway depends on the rate of the cleaved RCL insertion into the A β-sheet. If insertion of the RCL is prohibited, then cleavage of the complex

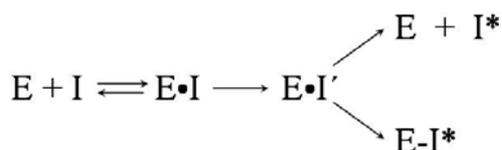


Figure 2. Reaction of a serine protease (E) with a serpin (I)

occurs, resulting in regeneration of the enzyme. Thus, the inhibitor has become a substrate.

Conformational changes: cleaved, "latent" and polymerized

It is known that many proteins have the inherent potential to undergo significant changes in their ordered structure. Serpins also have this potential for conformational change. It has been reported that there are at least four metastable conformational states of serpins: native, cleaved, "latent" and polymeric (Hekman and Loskutoff, 1985; Levin and Santell, 1987; Lomas et al, 1995a; Wardell et al, 1997; Dafforn et al, 1999; Zhou et al, 2001a).

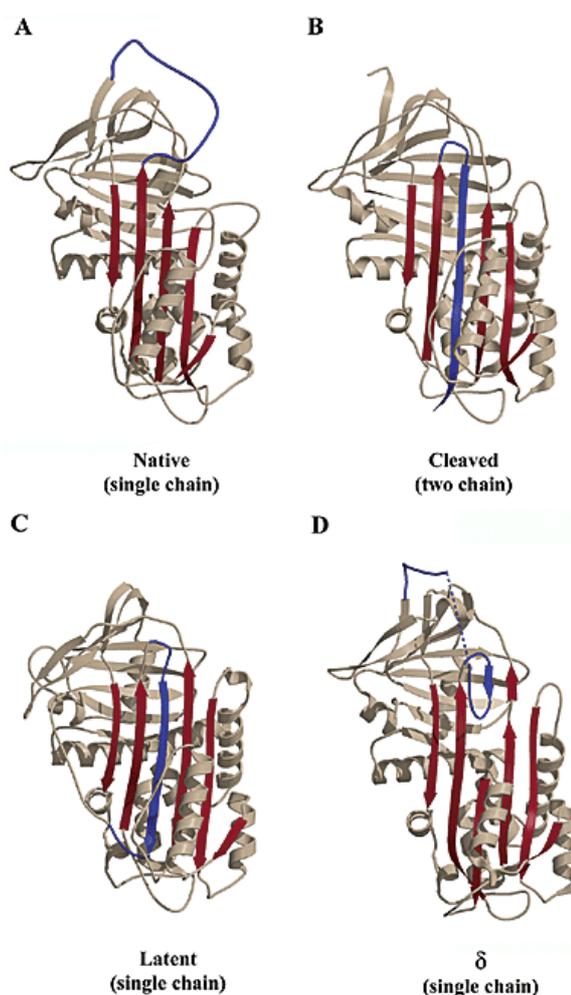


Figure 3. General structure of a Serpin (reproduced with permission, Lomas and coworkers, J Mol Biol 1998; 275: 419-425)

The conformation of the RCL in serpins shows a high variability, so there is no absolute structural requirement for its function. In the native non-inhibitory ovalbumin, the RCL is a 3-turn α -helix. In other serpins it has been found that the RCL typically has no regular secondary structure (Stein et al, 1990; Wei et al, 1994; Elliott et al, 1996, 1998; Li et al, 1999; Kim et al, 2001).

The structures of cleaved forms of inhibitory serpins are very stable and have therefore been quite easy to study by X-ray crystallography (Stein et al, 1991; Kaslik et al, 1997; Boudier and Beith, 2001). All of them show the same remarkable expansion of A β -sheet, due to the insertion of the RCL as a new strand into this β -sheet. During this energetically favourable process, the newly formed COOH-terminal and NH₂-terminal amino acids become separated by approximately 70 Å. This explains why the transition from uncleaved to cleaved serpin is practically irreversible.

It was found that PAI-1 produced by bovine endothelial cells to a large extent were inactive, but that it could become reactivated by treatment with strong denaturing agents (Hekman and Loskutoff, 1985). Therefore they called this form of PAI-1 for a "latent" molecule. Later it was found that PAI-1 indeed was synthesized as an active molecule, but that it due to stability problems at physiological conditions rapidly was converted to the "latent" form (Chmielewska et al, 1987). Structural analysis of "latent" PAI-1 revealed a conformational change similar to what has been found in the cleaved serpins, that is insertion of the RCL into the A β -sheet (Mottonen et al, 1992). A similar phenomenon was later reported regarding the heterodimeric structure of antithrombin (Carrell et al, 1994).

Lomas reported that also antitrypsin might be transformed into an inactive "latent" conformation by heating at high temperatures in stabilizing concentrations of sodium citrate (Lomas et al, 1995a). In this study, it has

been shown that also human antiplasmin might become converted into a "latent" form at low pH (Wang et al, 2004).

The ability of serpins to fold into a metastable state and subsequently undergo insertion of the RCL into the A β -sheet is an essential part of the protease inhibition mechanism. An unwanted consequence is the tendency of serpins to form polymers through RCL-sheet insertion mechanisms involving the loop of one molecule and an A β -sheet of another. This can be seen with most serpins by heating, followed by analysis of the products on a non-denaturing polyacryl amide gel (Paston et al, 1995). A ladder pattern is seen that corresponds to non-covalent dimers, trimers and higher order species. This polymerization can eventually also occur *in vivo*. The best known phenomenon of such loop-sheet polymers is the Z-variant of anti-trypsin (Lomas et al, 1992). This is a frequent natural variant in some populations in which Glu342 had been changed to a Lys (Jeppon and Laurell, 1988; Lomas et al, 1992; Elliott et al, 1998a). Zhou reported that recombinant native PAI-1 spontaneously forms polymers *in vitro* at low pH although with distinctly different electrophoretic patterns of polymerization (Zhou et al, 2001b).

Antiplasmin

Antiplasmin was found in 1976 by three different groups (Moroi and Aoki, 1976; Collen, 1976; Müllertz and Clemmensen, 1976). Antiplasmin is mainly produced by liver (Saito et al, 1982) and is reduced in liver diseases (Aoki and Yamanaka, 1978). It is a glycoprotein consisting of a single polypeptide with a molecular mass of 70 kd containing approximately 13% carbohydrates (Collen and Wiman, 1978). Holmes had determined the primary structure of mature human alpha 2-antiplasmin by DNA sequencing of overlapping cDNA fragments prepared from human liver mRNA. Mature human antiplasmin contains 452 amino acids as deduced from the cDNA sequence (Holmes et al, 1987) with the Asn in NH₂-terminal. Later on it had been found that antiplasmin

had an additional 12 amino acids at the NH₂-terminal and was composed of 464 amino acids with NH₂-terminal Met (Sumi et al, 1989; Bangert et al, 1993). In plasma Met-antiplasmin converts to Asn-antiplasmin (Koyama et al, 1994). There are four potential asparagine-linked glycosylation sites in the molecule, which comprises 452 amino acids. The plasma concentration of antiplasmin is 70 mg/L, approximately one-half the concentration of plasminogen on a molar basis. The half-life of the native inhibitor is approximately 3 days (Wiman and Collen, 1977). The gene of antiplasmin has been localized to chromosome 17p13 (Kato et al, 1993).

Antiplasmin has three functional properties. But its main physiological activity is inhibition of plasmin by forming an irreversible complex with plasmin (Wiman and Collen, 1978a, 1979). It inhibits plasmin extremely rapidly (Wiman and Collen, 1978a). In addition, it competes with fibrin regarding the affinity to plasmin(ogen), which is a very important regulatory phenomenon in the fibrinolytic process (Wiman and Collen, 1978c; Wiman et al, 1979a). Antiplasmin may also become cross-linked to the α -chains of fibrin during clotting (Aoki and Sakata, 1980; Reed et al, 1992).

Antiplasmin is capable of reacting with and forming stable enzymatically inactive complexes with plasmin. The reaction occurs in several steps (Wiman and Collen, 1978a, 1979; Wiman et al, 1978b; Christensen et al, 1996). The first, which has been proposed to be rate limiting, takes place between one of the lysine-binding sites in the plasmin molecule and a complementary site in the antiplasmin molecule. The second is a non-covalent interaction between the substrate-binding pocket in the plasmin active site and the scissile peptide bond in the RCL of antiplasmin. Subsequently, peptide bond cleavage occurs. After formation of an ester bond between antiplasmin and plasmin, major conformational changes occur both in the plasmin molecule and in the antiplasmin

molecule. The lysine-binding site mediated interaction has a major role in regulating the fibrinolytic system. The same site in the plasmin molecules are also involved in the interaction with fibrin. Fibrin-bound plasmin reacts much slower with antiplasmin as compared to free plasmin, thereby keeping the fibrinolytic process localized (Wiman and Collen, 1978a; Wiman et al, 1979a). It has been accepted that the COOH-terminal lysine (Lys452) in antiplasmin is responsible for the lysine-binding site mediated interactions with different proteins (Wiman and Collen, 1979; Wiman et al, 1979b, 1982; Sasaki et al, 1986; Holmes et al, 1987; Lijnen et al, 1987; Hortin et al, 1988; Sugiyama et al, 1988; Miles et al, 1991; Sjöström et al, 1997). In the present investigation it has been found that the COOH-terminal lysine is not at all important in this reaction, but that another site indeed is involved (see below).

Native antiplasmin is typically very stable at neutral pH, but at acidic pH or at elevated temperatures it rapidly becomes inactivated (Wiman and Collen, 1977). The reason for that antiplasmin becomes inactivated at decreased pH, or the mechanisms involved in these processes were not known at the start of the present project. However, now we know that both formation of "latent" molecules and polymerization processes are important parts of this phenomenon (see below).

PAI-1

PAI-1 was first found in conditioned medium from bovine aortic endothelial cells (Loskutoff et al, 1983). Almost simultaneously it was also found in plasma (Chmielewska et al, 1983; Kruithof et al, 1984). PAI-1 is the most important inhibitor of the plasminogen activators. In spite of that the concentration in plasma is low, it still is physiologically relevant, since it reacts extremely rapidly with both tPA and uPA (Chmielewska et al, 1988; Kruithof et al, 1986). The PAI-1 gene is located on chromosome 7q21.3-q22.3 and spans 12.2kb (Ginsburg et al, 1986; Ny et al, 1986; Bosma

et al, 1988). Increased plasma levels of PAI-1 are associated with the development of thrombotic diseases, both myocardial infarction (Hamsten et al, 1985, 1987; Paramo et al, 1985; Cortellaro et al, 1993) and deep vein thrombosis (Eriksson et al, 1989; Schulman and Wiman, 1996). Decreased levels of PAI-1 are connected with a mild bleeding tendency (Dieval et al, 1991; Carmeliet et al, 1993; Lee et al, 1993; Takahashi et al, 1996; Farrehi et al, 1998). In addition, data have been published suggesting that PAI-1 is able to modulate cell adhesion and cell migration (Pappot et al, 1995; Bajou et al, 1998).

Active PAI-1 in plasma is circulating in a complex with vitronectin (Declercq et al, 1988; Wiman et al, 1988; Sigurdardottir et al, 1990). PAI-1 binds to the NH₂-terminal somatomedin B domain in vitronectin (Seifert and Loskutoff, 1991; Sigurdardottir and Wiman, 1994; Lawrence et al, 1994; van Meijer et al, 1994). PAI-1 bound to vitronectin is more stable than free PAI-1 (Lindahl et al, 1989a). PAI-1 can also bind to heparin, which has been used for its purification (Lindahl and Wiman, 1989b). Heparin also changes the behaviour of PAI-1. Especially its reaction with thrombin becomes dramatically increased (Ehrlich et al, 1991; Keijer et al, 1991).

Three different conformations have been described for PAI-1, namely active, "latent" and substrate forms (Hekman and Loskutoff, 1985; Chmielewska et al, 1987; Declercq et al, 1992; Urano et al, 1992). At physiological conditions, active PAI-1 spontaneously changes its conformation and converts to the "latent" form with a half-life about 4 h (Wiman et al, 1988; Sigurdardottir and Wiman, 1990). PAI-1 is the only serpin to transform into the "latent" conformation spontaneously under physiological conditions. The crystal structure of "latent" PAI-1 revealed that the RCL is fully inserted into the A β -sheet (Mottonen et al, 1992; Aertgeerts et al, 1994). In the active conformation, the RCL is fully exposed on the surface of the molecule (Aertgeerts et al, 1994). In vitro, "latent"

PAI-1 can be reactivated by treatment with denaturing agents, followed by dialysis against a buffer without denaturing agents (Hekman and Loskutoff, 1985). Variants of PAI-1 that preferably works as substrates have been produced (Munch et al, 1993; Gils et al, 1996).

The active PAI-1 interacts rapidly both with tPA and uPA to form stoichiometric 1:1 inactive complexes (Lindahl et al, 1990). The inhibition of t-PA or u-PA by PAI-1 involves an interaction between the RCL of the serpin and the catalytic site of the protease. It has been suggested that also other types of secondary interactions are involved (Chmielewska et al 1988). Otherwise the reactions between PAI-1 and the plasminogen activators follow the same scheme as the other serpins (Lawrence et al, 1995; Wilczynska et al, 1995; Egelund et al, 1998).

Antithrombin

Antithrombin is a plasma protease inhibitor that may inactivate many active coagulation enzymes, but especially thrombin and Factor Xa. The concentration of antithrombin in human plasma is approximately 2 μ M (Murano, 1980). Decreased concentrations are connected with venous thromboembolism. The reaction rate between coagulation factors and antithrombin is typically quite low. However, in the presence of heparin the reaction rate is increased almost 1000-fold. This results in physiologically relevant reaction rates in the vicinity of healthy vessel wall endothelium in the microcirculation, due to the presence of heparane sulphate at the surface of endothelial cells. Antithrombin exhibits sequence, structural and functional homology to other members of the serpin family (Huber and Carrell, 1989). In addition, antithrombin has a defined binding site for heparin (Desai et al, 1998; Dementiev et al, 2004; Carter et al, 2005).

It has been found that during certain conditions antithrombin loses activity by polymerization processes (Corral et al, 2004). As with many other serpins this occurs by in-

serting the RCL from one molecule into the A β -sheet of another molecule. It has been reported that a hydrogen-bond between His334 (in strand 5 of the A β -sheet) and Asn186 (in strand 3 of the A β -sheet) is important in regulating this phenomenon. If this hydrogen-bond becomes disrupted under some conditions, such as increased temperature or decreased pH, polymerization will occur (Zhou et al, 2003).

Conformational disease

Recently, it has been realized that many diseases are caused by protein with decreased conformational stability (Carrell and Lomas, 1997, 2002; Kopito and Ron, 2000). This conformational instability may result in structures with less solubility, eventually leading to precipitation and intracellular accumulation of these proteins. Subsequently, this may lead to cellular damage. This type of phenomena might occur in many diseases, such as in Alzheimer's and Parkinson's disease (Tran and Miller, 1999; Walker and LeVine, 2000).

Regarding serpins it is known that the RCL from one molecule may be inserted into the A β -sheet of another serpin molecule, resulting in polymerization (Lomas et al, 1992; Stein and Carrell, 1995). One example of such a conformational disease is the relatively common condition of α_1 -protease inhibitor deficiency in humans (Berg and Eriksson, 1972; Brantly and Nukiwa, 1988; Brantly et al, 1988; Mahadeva et al, 1998, 1999). The main function of α_1 -protease inhibitor is to protect the tissues and against the neutrophil elastase. The lack of α_1 -protease inhibitor results in an imbalance between enzymes and inhibitors in the lung, which causes early-onset emphysema (Larsson, 1978). The Z mutation in α_1 -protease inhibitor is a glutamate to lysine substitution at residue P17. This amino acid is located at the head of strand five of the A β -sheet and at the base of the mobile RCL (Elliott et al, 1996). As a result of this mutation, the A β -sheet opens up and allows the RCL to become partially inserted into the A β -sheet.

The partial insertion in the top of the A β -sheet leaves the bottom-half open and ready to accept the RCL from another α_1 -protease inhibitor molecule (Lomas et al, 1993a,b; Gooptu et al, 2000; Sivasothy et al, 2000; Mahadeva et al, 2002). Sequential inter-molecular loop-sheet linkages occur with the formation of long chain polymers. The rate of forming polymers could be accelerated by raising the temperature and could be blocked by competing peptides (Lomas et al, 1992; Skinner et al, 1998). It was demonstrated that purified PAI-2 spontaneously polymerises at room temperature, similar to the Z-form of α_1 -protease inhibitor. The polymerization may lead to reduced secretion efficiency. The mechanism of PAI-2 polymerisation is inserting RCL into the A β -sheet of another molecule (Mikus and Ny, 1996; Wilczynska et al, 2003a, 2003b).

THE PRESENT INVESTIGATION

Aims of the study

- a. To increase the knowledge about the antiplasmin function in regulating the fibrinolytic system.
- b. To identify the complimentary site in antiplasmin to the lysine-binding site in plasmin(ogen).
- c. To understand the mechanisms involved in antiplasmin stability.

METHODOLOGY

Site-directed mutagenesis of antiplasmin cDNA

In vitro site-directed mutagenesis is a useful technique for studying protein structure-function relationships. QuikChange site-directed mutagenesis can easily be used to produce site-specific mutations in a double-stranded plasmid. This method does not need any subcloning mutagenesis of cDNA into vectors. In addition, this method does not require specialized vectors, restriction sites

or multiple transformations. The rapid four-step procedure generates mutants with more than 80% efficiency.

The oligonucleotide primers, each complementary to opposite strands of the plasmid, are extended during temperature cycling by PfuTurbo DNA polymerase. Following temperature cycling, the products are treated with Dpn I, which digests methylated and hemimethylated parental DNA template. The nicked DNA containing the desired mutations is then transformed into XL1-Blue super competent cells. After transformation, these cells repair the nicks in the mutated plasmid. Using the described technique we have constructed about 25 different antiplasmin variants. The nucleotide sequences of all the variants have been confirmed by DNA sequencing.

Expression of antiplasmin variants in S2 cells

The *Drosophila* Expression System (DES) utilizes a cell line derived from *Drosophila melanogaster*, Schneider 2 (S2) cells, and a simple plasmid vector for expression of heterologous proteins. S2 cells are easily maintained in loosely adherent or suspension cultures at room temperature and do not require CO₂. The vectors used for expression in S2 cells are very versatile, allowing inducible expression from the metallothionein (MT) promoter or constitutive expression from the actin Ac5 promoter. The expression can either be intracellular or secreted for simplified purification. Many native signal sequences are functional in S2 cells and can be used to secrete proteins. Stable cell lines expressing heterologous proteins can be generated in 3-4 weeks from a single cotransfection of the expression vector and pCOHygro selection vector.

To express the antiplasmin variants, transfected *Drosophila* Schneider S2 cells were cultured and extended in Schneider medium containing L-glutamine, heat-inactivated foetal bovine serum (final concentration 10%), 50U/ml penicillin G and 50µg/ml

streptomycin sulfate. After the volume had been expanded to 500 ml with a cell concentration of about $3 \times 10^9/L$, the cells were transferred to the same volume of *Drosophila* serum-free medium containing L-glutamine and penicillin/streptomycin at the same concentrations as above. Pluronic F68 (final concentration 0.05%) and CuSO₄ at a final concentration of 500 µM were also added. The cells were cultured with gentle stirring at room temperature in the darkness for 3 days. The cells were then removed by centrifugation at 2000 g for 30 min, and the supernatant containing antiplasmin was stored frozen at -70 °C.

Purification of antiplasmin variants

The native antiplasmin was purified from plasma as previously described, utilizing affinity chromatography on "kringles" 1-3 from plasminogen bound to Sepharose (Wiman, 1980).

Regarding purification of recombinant antiplasmin variants another method was developed, since we expected some of the variants to bind less good to the lysine-binding sites in plasminogen. Three steps were used for purification of the recombinant antiplasmin variants from the supernatant. First, the supernatant (about 500 ml) was mixed with DEAE-Sepharose CL6B in a batch procedure (about 25ml, equilibrated with 0.05 M Tris buffer, pH 8.0) during slow stirring for 2 hours at 4 °C. The suspension was then filtered through a Büchner-funnel and washed with about 1 l of 0.05 M Tris buffer, pH 8.0. The washed DEAE-Sepharose CL6B was packed in a column with a diameter of 2.5 cm and then washed with the same Tris buffer until the absorbance at 280 nm was less than 0.1. The elution was performed by a linear gradient from 0 to 0.4 M NaCl in the same Tris buffer. The antiplasmin concentration was determined in the different samples by ELISA. The antiplasmin-containing fractions were pooled and dialysed overnight against 0.04 M sodium phosphate buffer, pH 7.5, containing 0.1 M NaCl. The material was then applied to a Sephacryl S-300 HR

column equilibrated with 0.04 M sodium phosphate buffer, pH 7.5, containing 0.1 M NaCl. The fractions containing antiplasmin as determined by ELISA were pooled. The pool was subsequently applied to an anhydrotrypsin-agarose column (column volume about 2 ml), equilibrated with the same phosphate/NaCl buffer as above. The column was washed with the same buffer until the absorbance at 280 nm was less than 0.1. Elution was then performed with the equilibration buffer, also containing 0.3 M arginine-HCl. Fractions containing antiplasmin were dialysed against 0.04 M sodium phosphate buffer, pH 7.3, containing 0.1 M NaCl and then stored frozen at -70°C until use.

Determination of antiplasmin activity and antigen concentration

Antiplasmin activity was determined by a titration method against plasmin of known concentration, essentially as described (Wiman, 1981). Antiplasmin antigen concentration was determined by an ELISA method. For this purpose, Maxisorp microtiter plates were coated for 2 h at room temperature with goat anti-antiplasmin IgG, diluted in 0.1 M NaHCO_3 buffer, pH 9.6. The plates were then incubated at room temperature for 30 min with 0.04 M sodium phosphate buffer, pH 7.3, containing 0.1 M NaCl and 1 g/L BSA and washed three times with the phosphate/NaCl buffer, but without BSA. The samples (200 μL) to be analysed were compared with an antiplasmin standard (in the beginning native antiplasmin from human plasma was used, but later a wt-antiplasmin standard was prepared). The microtiter plates were incubated for 2 h at room temperature and then washed four times with the phosphate/NaCl buffer. Then, the anti-antiplasmin-IgG, conjugated with horseradish peroxidase, was added to the samples. After incubation for 1 h and washing the plates four times with phosphate/NaCl buffer, the horseradish peroxidase substrate o-phenylenediamine in the presence of H_2O_2 was added. After a new incubation for 10 minutes, 50 μL stop solution (3 M H_2SO_4)

was added to each well, and the A_{492} recorded in a microtiter plate reader.

Polyacrylamide gel electrophoresis (PAGE)

The non-denaturing PAGE system used is essentially that described by Jovin (1973). The separating gel contained 10 % (w/v) acrylamide in 0.38 M Tris-phosphate buffer, pH 8.8, whereas the stacking gel contained 4.5 % acrylamide in 0.12 M Tris-phosphate buffer, pH 6.9. The cathode buffer in the upper chamber contained 0.0529 M Tris buffer, pH 8.2, containing 0.0684 M glycine, while the anodic buffer in the lower chamber contained 0.1 M Tris buffer, pH 7.8. All the gel electrophoresis was performed in Bio-Rad Mini Protean II apparatus at room temperature. SDS/PAGE was performed in a Mini-protean II electrophoresis apparatus (Bio-Rad, Stockholm, Sweden) as described by Laemmli (1970). Proteins were separated in 10 % (w/v) polyacrylamide gels and stained with Coomassie Brilliant Blue R-250.

Determination of rate constants in the reaction between plasmin and the antiplasmin variants

In order to study the reaction between plasmin and the different antiplasmin variants, the two reactants were mixed in 0.1 M sodium phosphate buffer, pH 7.3, or the same buffer containing 1.0 mM 6-aminohexanoic acid. The final plasmin concentration (active site titrated) used in these experiments was 0.6 nM, whereas the antiplasmin concentration varied between 1 and 5 nM. After specified times of reaction (0–300 s), samples were withdrawn into tubes containing 0.6 mM Flavigen Pli (final concentration), 20 mM 6-aminohexanoic acid and polyclonal rabbit anti-human antiplasmin IgG (1 mg/mL). By this procedure further inhibition of plasmin was rapidly and efficiently decreased, allowing long incubation times with the plasmin substrate, which is necessary to accurately measure the low plasmin concentrations. After incubation for 1.5 h, plasmin cleavage of the chromogenic substrate was

stopped by addition of acetic acid (final concentration 1%, v/v) and the A_{405} recorded. A_{405} is thus a reliable measure of the residual plasmin concentration at the time of sampling. Then, the reaction rate constants were calculated from the results using the classic formula for second-order reactions, using data obtained before 50% of the plasmin was inhibited. In the experiments performed in the presence of 6-aminohexanoic acids, in which the antiplasmin concentration was almost 10-fold higher than the plasmin concentration, pseudo-first-order conditions were assumed and the reaction rate constants were calculated from the half-lives of plasmin in these experiments (also before 50% of the plasmin activity was inhibited).

Binding of antiplasmin variants to plasminogen or “kringles” 1-3 measured by surface plasmon resonance

The interaction between plasminogen or “kringles” 1-3 in plasminogen and each one of the antiplasmin variants were analysed by surface plasmon resonance in a BIAcore 2000 (BIAcore AB, Uppsala, Sweden). Plasminogen or “kringles” 1-3 was immobilised at the carboxymethyl surface of a CM5 chip (BIAcore AB) using amine coupling according to the manufacturers instructions. For the immobilization procedure, the proteins were diluted to 2.5 $\mu\text{g/ml}$ in an acetate buffer, pH 5.5 (BIAcore AB) and coupled to the surface in one of the flow cells, at a flow rate of 10 $\mu\text{l/min}$ for 7 minutes. At the same time, a reference surface was prepared in one of the other flow cells. For the kinetic measurements, 5 different concentrations of each antiplasmin variant were injected for 3 minutes followed by a 10 minutes dissociation phase. At the end of each cycle, the surface was regenerated by a one-minute injection of 1 M NaCl, followed by a one-minute injection of 20 mM 6-aminohexanoic acid. The experiments were carried out in 40 mM phosphate buffer, pH 7.5, containing 0.1 M NaCl, at a flow rate of 30 $\mu\text{l/min}$. Assuming a 1:1 stoichiometry in a bimolecular reaction model, the association and dissociation phases of the obtained sensorgrams were analysed by curve fitting, us-

ing the BIAevaluation software (version 3.1). The experimental procedure with wt-antiplasmin was carried out at several different occasions, during quite a long period of time. Deviating results were obtained, if analyses were performed several days after the coupling procedure. For this reason, in the experiments where all the different antiplasmin variants (including wt-antiplasmin) were included, the chips were used within 24 hours after coupling.

Computer modelling of the COOH-terminal 25 or 40 amino acid residues in antiplasmin

A computer model of the C-terminal 25 or 40 amino acids in antiplasmin was constructed by CS Chem3d Ultra, version 7.0 (Cambridge Soft, Cambridge, MA, USA), followed by energy minimization utilizing the MM2 protocol. Modelling was performed on different lengths of the C-terminal portion of the antiplasmin molecule, ranging from 25 to 50 residues from Lys452. However, energy minimization did not work well on structures with more than 40 amino acids.

RESULTS AND DISCUSSION

The site in antiplasmin interacting with lysine-binding sites in plasmin (Papers I and II)

The reactions between plasmin and different antiplasmin variants

In order to study the lysine-binding-site mediated interaction between plasmin and antiplasmin and possibly identify the amino acids in antiplasmin of importance for this interaction we constructed 7 different single site mutants in COOH-terminal portion of antiplasmin. Two mutants were produced in which the COOH-terminal lysine (Lys452) was replaced by amino acids without a positive charge (Glu or Thr). We also produced variants in which other charged amino acids in

this portion of the molecule were changed to either uncharged residues or residues of opposite charge (Figure 4). All the antiplasmin variants were expressed in insect cells (*Drosophila* S2 cells), purified, and characterized with regard to their reactions with plasmin. The rate constants for the reactions between plasmin and the antiplasmin variants, in the absence or presence, of 6-aminohexanoic acids, are demonstrated in table 1. The rate constants for the reactions between ‘native’ human antiplasmin and plasmin in the presence or absence of 6-aminohexanoic acids are also shown for comparison.

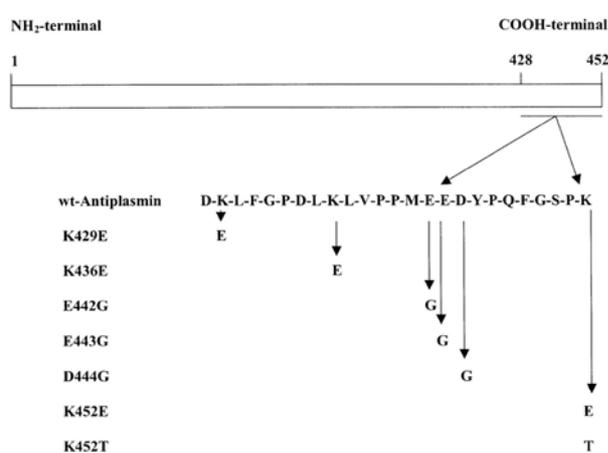


Figure 4. Mutations in the antiplasmin COOH-terminal portion

All variants of antiplasmin except for K436E had a rate constant higher than $10^7 \text{ m}^{-1}\cdot\text{s}^{-1}$. This is not far from the rate constant obtained with native antiplasmin. In addition, the method used here gave very similar results regarding the rate constants as those reported earlier (Wiman and Collen, 1978a; Christensen et al, 1996). Interestingly, the mutants K452E and K452T did not differ in behaviour from wt-antiplasmin, suggesting that the COOH-terminal lysine is of little importance in the lysine-binding-site mediated interaction between plasmin and antiplasmin. On the other hand, the variant K436E reacted much more slowly (about five fold) than the other variants, rather suggesting that this residue is important.

In the presence of 6-aminohexanoic acids, the reaction rate decreased 10-fold or more for most variants. Also in this case the results

with wt-antiplasmin and the mutants K452E and K452T did not differ. Only the variant K436E was less affected by 6-aminohexanoic acids (2.5-fold decrease in reaction rate), again suggesting that this residue is involved in the lysine-binding-site mediated interaction between plasmin and antiplasmin.

The Interaction between Plasminogen and Antiplasmin Variants as Studied by Surface Plasmon Resonance

To further elucidate this process we studied the interaction between these antiplasmin variants and immobilized plasminogen or an elastase fragment from plasminogen constituting “kringles” 1-3, using surface plasmon resonance. The antiplasmin variants studied were wt-antiplasmin, K429E, K436E, E443G, D444G, K452E and K452T. Clearly, the two mutants K452E and K452T reacted quite similarly to wt-antiplasmin. The K_D is in fact quite similar for all these variants, again suggesting that Lys 452 is not involved in the lysine-binding site interaction between plasminogen and antiplasmin. On the other hand the mutant K436E displays a much lower association rate constant, both to intact plasminogen and to “kringles” 1-3. In fact, the affinity was very poor, making it difficult to get reliable values for the K_D using the present conditions. Nevertheless, the K_D values for the interaction between the K436E mutant and either plasminogen or the “kringles” 1-3 fragment were estimated as about 40-fold higher as compared to wt-antiplasmin. Thus, our data show that K436 is indeed very important in the lysine-binding site interaction between plasminogen and antiplasmin. Taken together with our previous investigation, using completely different technology, our data proves that Lys436 but not Lys452 in antiplasmin is responsible for the interaction through the lysin-binding sites in plasmin. The mutant E443G displays somewhat lower affinity for both plasminogen and “kringles” 1-3, as compared to wt-antiplasmin. It is possible that this residue is also involved in the lysine-binding-site mediated interaction.

Table 1. Rate constants (in $10^6 \text{ M}^{-1}\text{s}^{-1}$) in the reactions between plasmin and the different antiplasmin variants in the absence (No 6-AHA) or the presence (6-AHA) of 1.0 mM 6-AHA

Variant	No 6-AHA	6-AHA
Native	25.3±1.7	2.5
Wt-AP	10.9±0.3	1.1
K429E	27.3±2.5	2.7
K436E	2.1±0.3	0.8
E442G	19.5±1.0	1.5
E443G	24.3±1.2	1.6
D444G	21.6±0.9	1.6
K452E	11.5±0.7	0.8
K452T	12.7±1.0	0.9

Molecular modelling of the COOH-terminal portion of antiplasmin

The amino-acid sequence of the COOH-terminal 40 residues in antiplasmin is GNKD-

FLQSLKGFPRGDKLFGPDLKLVPPMEE-DYPQFGSPK-OH (Holmes et al, 1987). The COOH-terminal lysine is residue 452 in the antiplasmin molecule. Molecular modelling resulted in the structures shown in figure 5. We constructed a number of models with different lengths, ranging from 25 to 50 residues from the COOH-terminal Lys452. All models were similar around the two sites Lys452 and Lys436. In these models, the side chain of the COOH-terminal lysine residue (K452) seems to be in the close vicinity to the side chain of Phe448, perhaps explaining why it is not involved in the lysine-binding-site mediated interaction. Lys436, on the other hand, is clearly exposed at the surface of the molecule. In addition the negatively charged side-chain of Glu443 is within the expected range from the positively charged side-chain of Lys436 to form a potentially complementary site to a

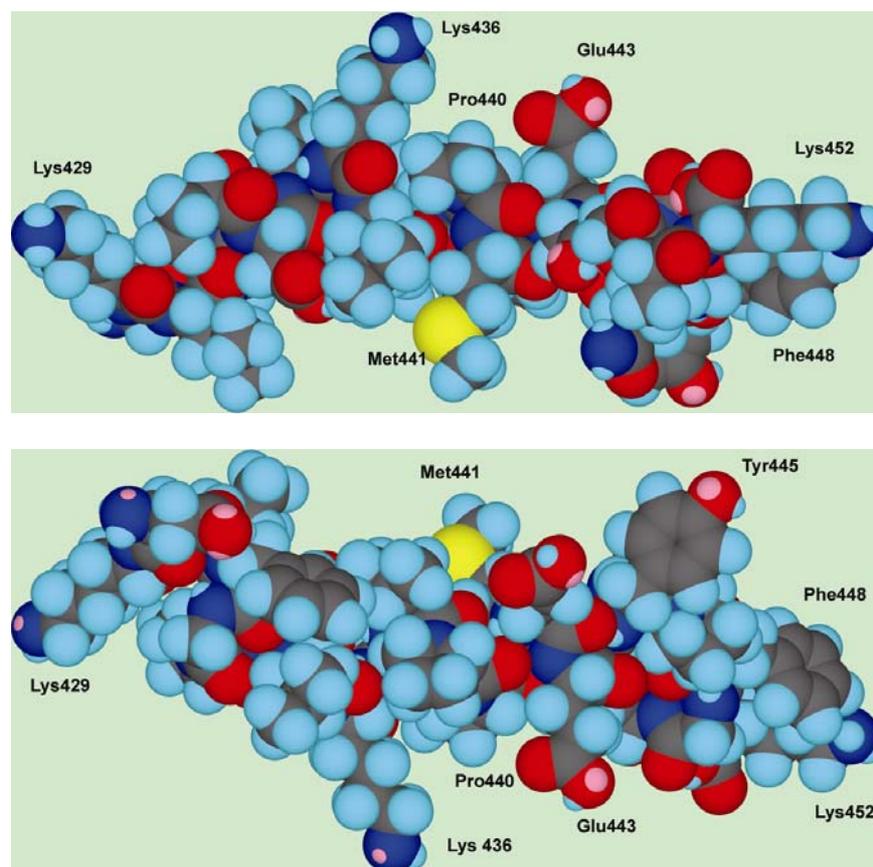


Figure 5. Computer model of the COOH-terminal 25 amino acids in antiplasmin viewed from two different angles. Some of the residues are numbered to facilitate viewing.

lysine-binding site. Also these two charged side-chains seem to be separated by the hydrophobic part of the side chain in Pro439.

Antiplasmin stability (Papers III and IV)

Inactivation of antiplasmin

Previously, it has been demonstrated that human antiplasmin is quite stable at physiological conditions, but that activity is rapidly lost at acidic conditions (Wiman, 1977). Here we have studied that process in more detail. Antiplasmin was purified from plasma as previously described (Wiman, 1980, 1981) using affinity chromatography on "kringles" 1-3 in plasminogen, insolubilised to Sepharose 4B, followed by gel filtration on Sephacryl S-200. The antiplasmin preparation appeared as a single band on SDS/PAGE, with an estimated molecular weight of about 70,000. It was almost completely active as judged from activity measurements and from its ability to form a SDS stable complex with plasmin.

The purified antiplasmin was inactivated at decreased pH (pH 4.9-6.1) and at different temperatures (22 and 37 °C). The decline in activity typically followed first-order kinetics. After plotting the logarithm of antiplasmin activity against time, the half-lives were calculated. In addition to this, we also performed inactivation experiments at elevated temperatures (45 and 55 °C), but only at pH 5.5. The results are summarised in table 2. Almost no inactivation occurred at pH above 6.1 at temperatures below 37 °C. However, at elevated temperatures (above 45 °C), a rapid inactivation was observed. In order to further study the mechanism for inactivation of antiplasmin at different pH and at different temperatures, we also followed these processes by PAGE at non-denaturing conditions. The results of the PAGE were also analysed by running Western blots, in order to confirm that the "polymerized" bands seen really constituted antiplasmin. At elevated temperatures, our data indicated that polymerization rapidly occurred, which most likely is responsible for the rapid inactivation under these conditions. In con-

trast, inactivation of antiplasmin at pH 5.5 and 22 °C occurred mostly without any visible polymerization tendency. Thus the mechanism for inactivation at normal temperatures is different than that at elevated temperatures. One possible explanation for this is the formation of "latent" molecules. Our data have demonstrated that inactivation of antiplasmin at slightly decreased pH is a quite complicated process, involving both polymerisation and perhaps also formation of "latent" molecules.

Reactivation of antiplasmin

Antiplasmin, which has been inactivated at pH 4.9 for 24 h at 22 °C, does not have any measurable activity. If such material was incubated at neutral pH, by dilution with 0.1 M sodium phosphate buffer, pH 7.3, a gradual increase in antiplasmin activity was observed. The maximal activity obtained was still low, about 5% of the theoretical value. Antiplasmin, inactivated in the same way as above was also treated with different concentrations of guanidinium chloride (1-4 M), followed by dialysis and measurement of antiplasmin activity. In this way much higher antiplasmin activity was obtained. The maximal activity obtained corresponds to a specific activity of about 37% during 0.5-2 h incubation with 4 M guanidinium chloride. Antiplasmin, inactivated at pH 6.2 at 55 °C, which mainly yielded polymerized material, was incubated at 3.2 M guanidinium chloride, followed by dialysis against 0.1 M sodium phosphate buffer, pH 7.3. In this case, activity measurements revealed that very little antiplasmin activity was recovered (<3%).

Structures of importance for the stability of antiplasmin as studied by site-directed mutagenesis

We decide to study the inactivation of antiplasmin at decreased pH or slightly increased temperatures by investigating the stability of single-site mutants of antiplasmin. The rationale for choosing the specific variants were as follows: The amino acid sequences of PAI-1 and antiplasmin were compared, since PAI-1 has a reversed pH-dependent stability as

compared to antipiasmin (Hekman and Loskutoff, 1985; Lindahl et al, 1989a; Lawrence et al, 1994; Mångs et al, 2000; Wang et al, 2004).

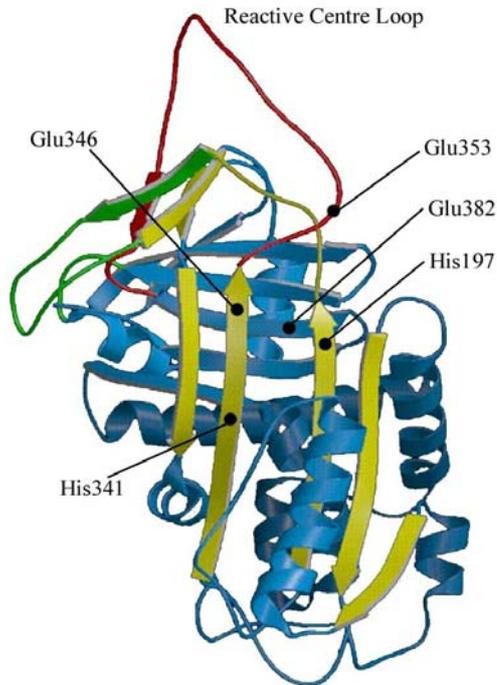


Figure 6. A schematic picture of PAI-1 (adapted from Aertgeerts *et al* 1994, *J.Struct.Biol.*, 113:239-245), with the most important mutations marked. Since antipiasmin has not been crystallized, we rather show a model of PAI-1.

Indeed, this revealed some important sequence differences in some of the interesting parts in antipiasmin, which might be involved in the stability mechanism (RCL, the A β -sheet or the B β -sheet). Some of these differences may be the reason for why PAI-1 and antipiasmin are stable at different pH. With these differences in mind, we used site-directed mutagenesis to produce 11 single-site mutants (His197Thr; Gln212His; Ile246His; His250Asp; His250Thr; His266Glu; His341Asp; His341Thr; Glu346Thr; Glu353Ala; Glu382His) within these regions. The inactivation of wt-antipiasmin was studied at different pH (pH 4.7-5.6) at room temperature and the decline in activity was found to follow first-order kinetics. The lower the pH, the faster inactivation occurred. The half-life of wt-antipiasmin is very similar to that of “native” antipiasmin at pH 4.9 (Wang et al, 2004). The half-lives of 4 of these antipiasmin variants (His250Asp; His250Thr; His266Glu; His341Asp) did not deviate much from the result obtained with wt-antipiasmin. Two of the mutants (Gln212His and Ile246His) had a slight increased half-life. However, four of the mutants (His197Thr, His341Thr, Glu353Ala and Glu382His) displayed a significantly increased stability. The mutant His341Thr was found to have the longest half-life at pH 4.9, about 7 fold increased stability as compared to wt-antipiasmin. One mutant, Glu346Thr, had a shorter half-life than wt-antipiasmin.

Table 2. Inactivation of antipiasmin as a function of pH and temperature with time. The decline in antipiasmin activity followed first-order kinetics. The data shown (mean of three independent experiments) represent the half-lives in the transition from active to inactive inhibitor.

pH	22 °C	37 °C	45 °C	55 °C
4.9	14 min	< 5 min		
5.2	47 min	9 min		
5.5	24 h	9 h	~1 h	~10 min
5.8	> 30 h	27 h		
6.1	> 30 h	> 30 h		

The mutants (His197Thr; His341Thr; Glu346Thr; Glu353Ala; Glu382His) (Figure 6), which have deviating properties as compared to wt-antiplasmin, have been studied more extensively. These variants were also analysed by PAGE under non-denaturing conditions after incubation both at pH 4.9 at room temperature for 2 hours, and at 45 °C at pH 5.5, for 2 hours. After incubation at pH 4.9 for 2 hours at room temperature, it was obvious that several of the variants had polymerised to different extent. Interestingly, Glu346Thr, which has a shorter activity half-life than wt-antiplasmin, only have a very small tendency to polymerise under these conditions. Therefore, inactivation of this mutant must occur by other means than polymerisation, e.g. by formation of “latent” molecules. Also, regarding the mutants His197Thr and Glu382His, the results are in favor of formation of “latent” molecules, as the main pathway of inactivation. Treatment of the antiplasmin variants at slightly increased temperature (45 °C) at pH 5.5 for 2 hours did not display any major differences on PAGE, as compared to the untreated samples, in spite of that the activity had decreased to less than 40% (10-40%). Even if the decline in antiplasmin activity at decreased pH is similar, it seems that the recombinant wt-antiplasmin is more prone to

polymerise as compared to the “native” antiplasmin. On the other hand at elevated temperature wt-antiplasmin rather transforms into a “latent” conformation, which is different from the “native” antiplasmin. This is a clear difference in behaviour between “native” antiplasmin and recombinant wt-antiplasmin.

Table 3. Inactivation of antiplasmin variants at pH 4.9 and 22 °C. The decline in antiplasmin activity typically followed first-order kinetics. The data shown represent the half-lives in the transition from active to inactive inhibitor. The result for “native” antiplasmin is included for comparison.

Variants	half-life (min)
“native”	14.0
wt-AP	12.4
His197Thr	38.9
Gln212His	20.5
Ile246His	20.5
His250Asp	14.0
His250Thr	11.2
His266Glu	10.0
His341Asp	11.8
His341Thr	99.0
Glu346Thr	6.5
Glu353Ala	24.2
Glu382His	23.7

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